2018 Long-Term Stewardship Conference

Performance of an In Situ Hydroxyapatite Permeable Reactive Barrier at the Old Rifle Uranium Processing Mill Site

Mark J. Rigali Sandia National Laboratories

2.1 Groundwater Compliance Challenges

Sandia National Laboratories is a mult-imission laboratory managed and operated by National Technology and Engineering Solutions of Sandia LLC, a wholly owned subsidiary of Honeywell International Inc. for the U.S.

Department of Energy's National Nuclear Security Administration under contract DE-NAO00352-5.

SAND2018-7026 PE

Kenneth H. Williams

Robert C. Moore

Lawrence Berkley National Laboratories

Pacific Northwest National Laboratories

James Szecsody

Jon R. Luellen

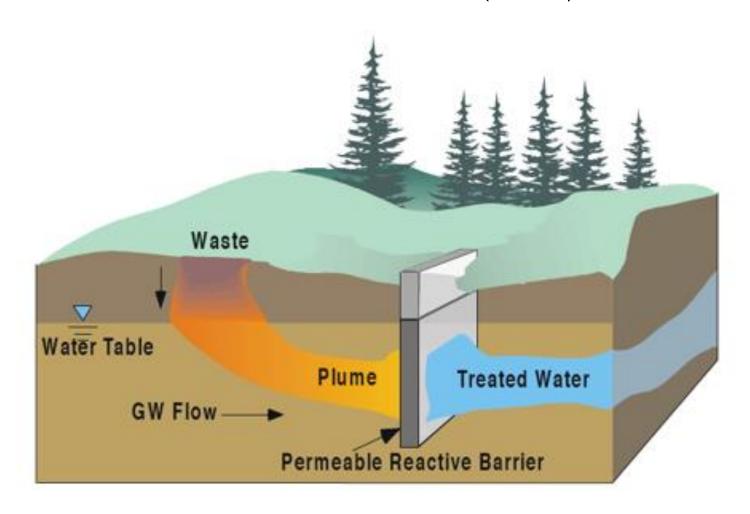
Pacific Northwest National Laboratories

AECOM

Presentation Outline

- Overview of Apatite Permeable Reactive Barriers (PRBs)
- The Old Rifle Site
- Deployment of the Apatite PRB at the Old Rifle Uranium Mill Site
- Performance of the Apatite PRB for Uranium Remediation
- Summary

Permeable Reactive Barrier (PRB) Technology



Conventional PRB Construction





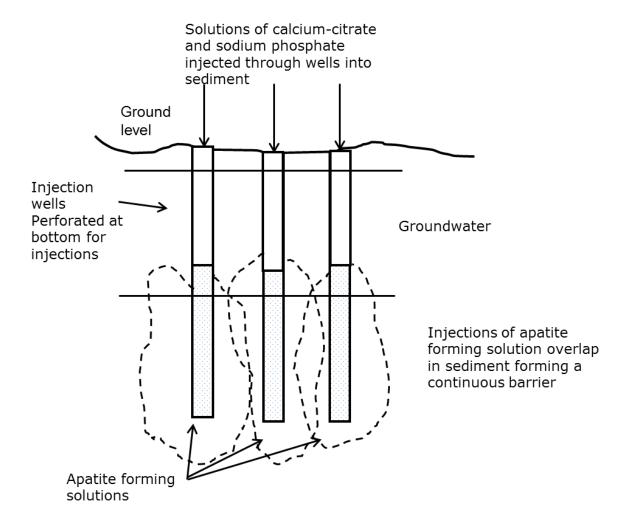
Trenching Followed by Backfilling with a Reactive Media

Jet or Grout Injection of A Reactive Media

Apatite as a PRB Material

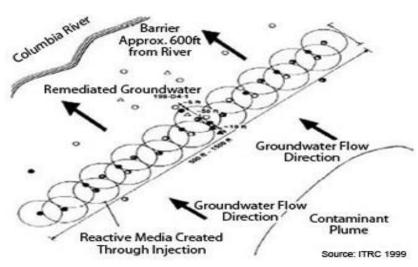
- Composed of calcium phosphate, $Ca_{10}(PO_4)_6(OH)_2$
- Common mineral in the environment and as a main component of bone and teeth.
- Very stable in the environment, extremely low solubility.
- Can be used to sequester a wide variety of radionuclides, heavy metals and other contaminants through substitution into the structure or sorption onto the surface as metal phosphate compounds.
- The high affinity for these contaminants, stability and very low solubility make apatite an ideal sorbent for immobilization of many contaminants.
- Can be formed <u>in situ</u> by solution injection in the subsurface...

Forming an Apatite PRB

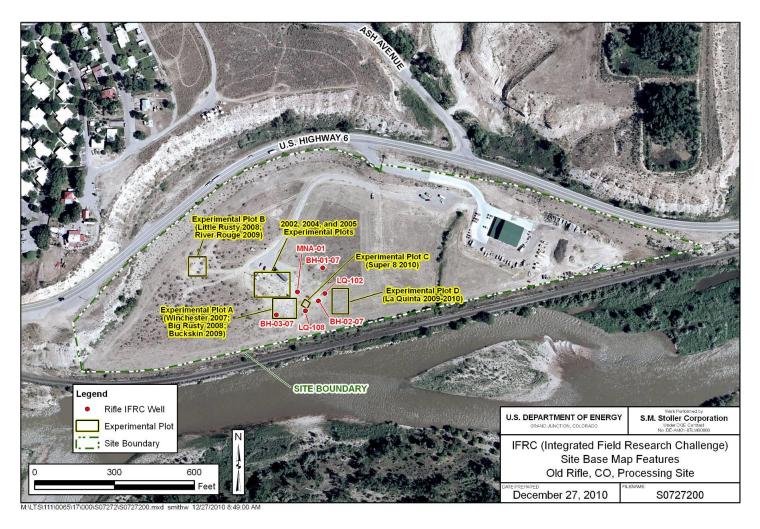


Successfully Deployed at the Hanford Site





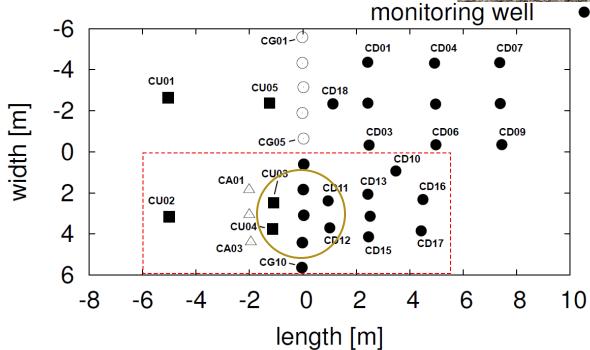
Old Rifle Site IFRC Experimental Plots



Plot C Wells - Apatite Injection Experiment

CG-08 CU-04 CA-02

flow direction



Field Deployment of Apatite Forming Solutions

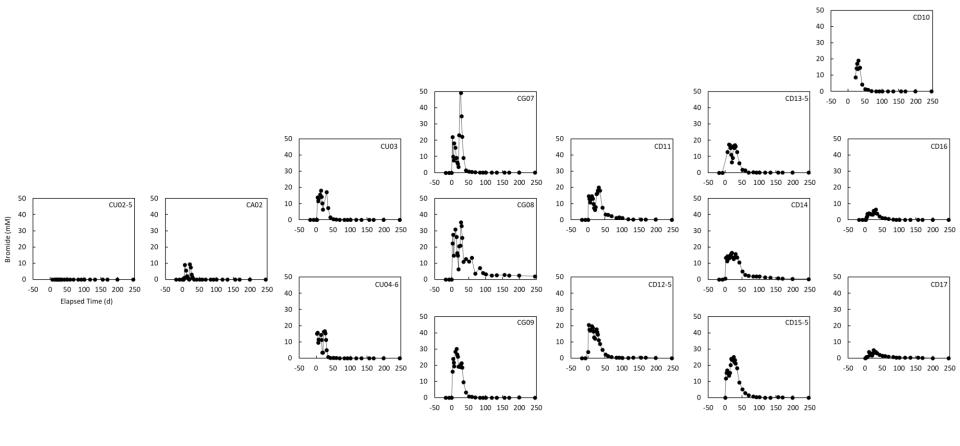
- Injection Formulation:
 - Tank A: Solution containing 40mM of Ca²⁺,100mM citrate and 40mM Bromide
 - Tank B: solution containing 40mM phosphate
- Injection of 4,800 gallons of solution total (~ 5 pore volumes)
- Three injections total. (Nov. 14th, 21st, and Dec. 5th, 2017).
- Injection Rate Varied: ~4 to 16 liters per minute.
- Injections completed within 24 to 56 hours.
- Maintained tank temperature at 20-25 C.
- Primary injection well is CG-08.

Field Deployment of Apatite Forming Solutions

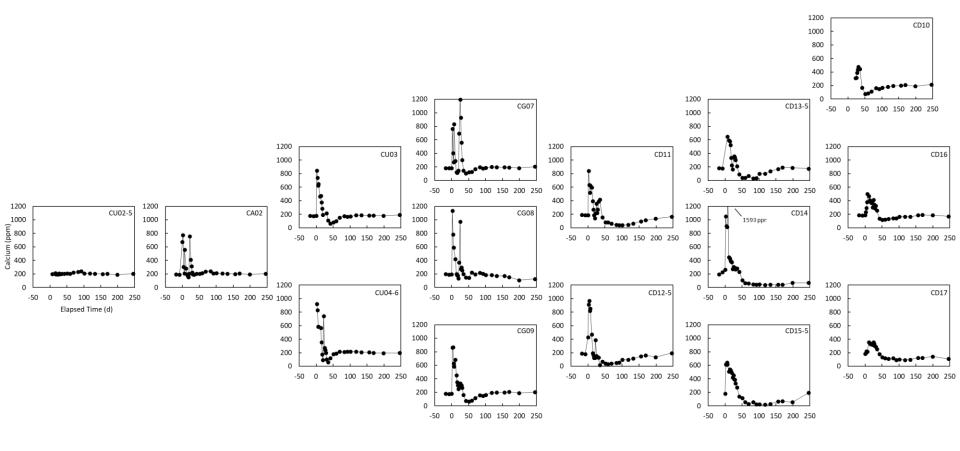


2018 LTS Conference

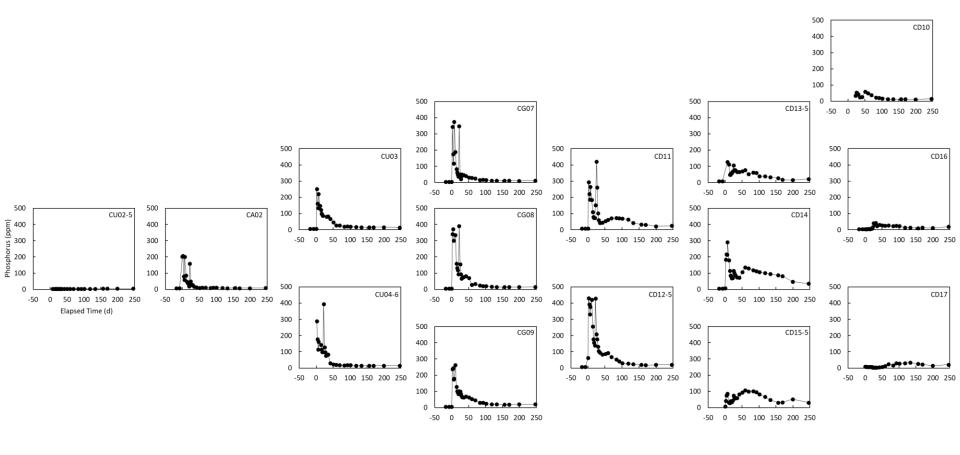
Bromide Concentrations (ppm)



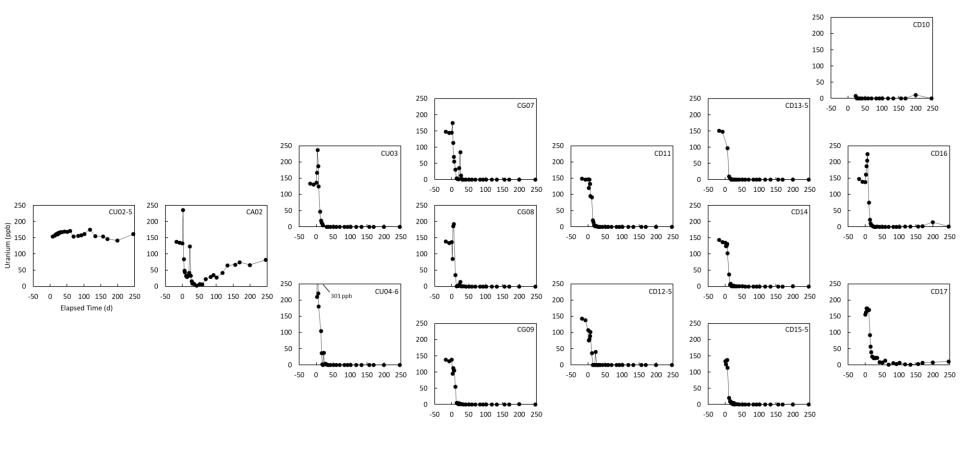
Calcium Concentrations (ppm)



Phosphorous Concentrations (ppm)



Uranium Concentrations (ppb)



Summary

- Apatite is an effective sorbent for a variety of radionuclides and heavy metals.
- It can be deployed as a Permeable Reactive Barrier (PRB) by injecting simple aqueous-based solutions directly into the subsurface.
- The apatite PRB field experiment indicates very effective uptake of uranium after more than 200 days of monitoring.
- Low cost deployment as compared to traditional PRB deployments that is also minimally invasive to the site.